

## Collision Theory Pogil Answer Key

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### Collision Theory Pogil Answer Key

Collision Theory Key Questions 1. Did a reaction take place between Reactant A and Reactant B in Scenario 1? Why or why not? Explain your reasoning in terms of the nature of the collision. 2. Did a reaction take place between Reactant A and Reactant B in Scenario 2? Why or why not? Explain your reasoning in terms of the nature of the collision. 3.

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Collision Theory . Key Questions . 1. Did a reaction take place between . Reactant A (baseball bat) and . Reactant B (baseball) in Scenario 1? Why or why not? Explain your reasoning in terms of the . nature . of the collision. 2. Did a reaction take place between . Reactant A. and . Reactant B. in Scenario 2? Why or why not? Explain your reasoning in terms of the . nature

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## **Collision Theory - Impact for a Chemical Reaction**

Collision Theory Model: Collision Theory In the picture below, the baseball bat represents Reactant A and the baseball represents Reactant B. A reaction will only be successful if the batter hits a homerun. If the batter does not hit a homerun, the reaction will be considered a failure. Now, read the four scenarios below and answer the key ...

## **Collision Theory - Impact for a Chemical Reaction**

Model: Collision Theory In the picture below, the baseball bat represents Reactant A and the baseball represents Reactant B. A reaction will only be successful if the batter hits a homerun. !!!If the batter does not hit a homerun, the reaction will be considered a failure. Now, read the four scenarios below and answer the key questions that follow. ...

## **01\_Collision\_POGIL.docx - Model Collision Theory In the ...**

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Collision Theory Key Questions 1. Did a reaction take place between Reactant A and Reactant B in Scenario 1? Why or why not? Explain your reasoning in terms of the nature of the collision. 2. Did a reaction take place between Reactant A and Reactant B in Scenario 2? Why or why not? Explain your reasoning in terms of the nature of the collision. NO

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1) Collision Theory POGIL . 2) Factors the Affect Rate . 3) Potential Energy Diagrams . 4) Dynamic Equilibrium POGIL . 5) LeChatelier's Principle POGIL . 6) LeChatelier's Principle . 7) Equilibrium Constants . 8) ICE Problems . 9) Ka and Kb Problems

## **AHS Chemistry Resource Site - Unit 6 - Rates & Equilibrium**

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MAIN Idea Collision theory is the key to understanding why some reactions are faster than others. 16.2 Factors Affecting Reaction Rates MAIN Idea Factors such as reactivity, concentration, temperature, surface area, and catalysts affect the rate of a chemical reaction. 16.3 Reaction Rate Laws MAIN Idea The reaction rate law is an experimentally determined

## Chapter 16: Reaction Rates

Kinetic Molecular Theory Key Questions 1. What causes a gas to exert pressure when confined in a container? of 2. How does the total volume of gas particles compare to the volume of the space ... ©POGIL - 2005 4/4 Authored by ... The motion of a gas molecule is usually stopped by collision with two different things. Based on the diagram 3. 4 ...

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Collision Theory. According to the collision theory, “the molecules of reactants are assumed to be hard spheres and the reactions are assumed to occur only when these spheres (molecules) collide with each other”. So it was important to quantify the number of collisions occurring in order to form products so that we can have a clear picture of the reaction, and hence came the term collision frequency.

## Collision Theory - Definition & Explanation, Arrhenius ...

5. Based on your responses to Key Questions 1-4 and your reasoning, what insight has your team gained about the term effective collision? -D 6. Based on your answer to Key Question 5, complete the following statement: Collision theory states that a reaction is most likely to occur if and OPOGIL 2005, 2006

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